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Model 714 NO₂/NO/O₃ Calibration Source

Calibrate 3 Gases Using One Convenient Instrument



The Model 714 NO₂/NO/O₃ Calibration Source^M combines our Model 408 Nitric Oxide Calibration Source^M with our Model 306 Ozone Calibration Source^M to provide a highly portable transfer standard for three pollutant gases without the need for a compressed cylinder of nitric oxide. Calibrated concentrations of O₃ are produced by photolysis of oxygen in scrubbed ambient air, and calibrated concentrations of NO are produced by photolysis of nitrous oxide, N₂O, via a patented process,

$$N_2O + h\nu \rightarrow N_2 + O^*$$

 $O^* + N_2O \rightarrow 2 NO$

where O^{*} is electronically excited oxygen atoms, $O({}^{1}D_{2})$ (<u>U.S. Patent No. 10,207,927</u>). In addition, calibrated concentrations of NO₂ are produced by gas phase titration (GPT) of NO with O₃ using the same method as

conventional GPT NO/NO₂ calibrators. The concentrations of all three pollutant gases are directly traceable to the NIST photometric standard for ozone – eliminating uncertainties associated with the stability of nitric oxide standards.

Nitrous oxide is provided by disposable 8 or 16 oz cartridges typically used for making whipped cream, eliminating the requirement for a compressed gas cylinder and thereby enhancing safety and portability. The Model 714 is designed to accompany sales of our new FEM Model 405 nm NO₂/NO/NOx Monitor and replace current calibration systems for all ozone, NO and NO₂ monitors. The Model 714, which has an intuitive touch screen user interface, is provided in a rack mount-sized instrument case matching that of the Model 405 nm.



Small and inexpensive N₂O cartridges for your calibrations!

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Specifications: Model 714 NO₂/NO/O₃ Calibration Source

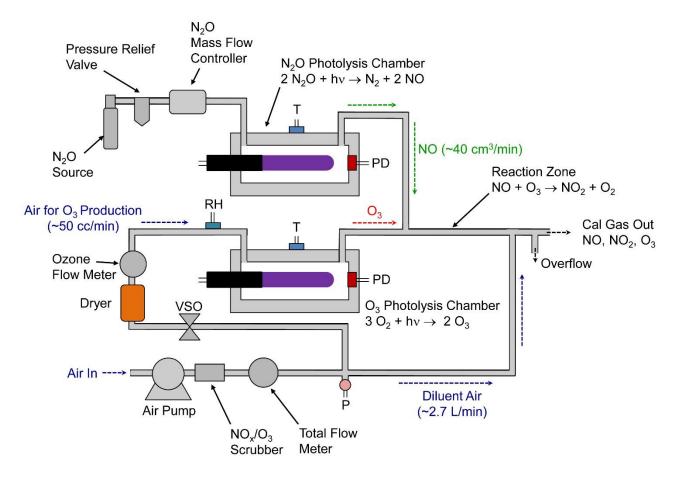
| Method of NO Production | UV photolysis of Nitrous Oxide (N ₂ O) at 185 nm |
|---|---|
| Method of O ₃ Production | UV photolysis of Oxygen (O ₂) at 185 nm |
| Method of NO ₂ Production | Reaction of O_3 with excess NO (gas-phase titration, GPT) |
| NO ₂ Concentration Range | 0-500 parts per billion by volume (ppb) |
| O ₃ Concentration Range | 0-500 ppb |
| NO Concentration Range | 0-1000 ppb |
| Output Flow Rate | ~2.0 to 2.7 L/min volumetric (2.5-2.7 typical) |
| Precision and Accuracy | Greater of 2.0 ppb or 2.0% of NO or O_3 concentration |
| Response Time for Change in Calibrant Gas Output Concentration | < 30 s to reach 95% of selected concentration |
| Input and Output | Touch Screen LCD Display |
| Dimensions | Rackmount: 17" w × 14.5" d × 5.5" h (43 × 37 × 14 cm) |
| Weight | 15.5 lb (7.0 kg) |
| Power Requirements | 120/240 V AC with 5-amp power pack, or 12V DC; 1.5 A at 12V,18 watt (operation) Max: 3.5 A at 12V, 42 watt (warmup) |
| Programmable Calibrated Gas Output | Yes, up to 99 different repeatable sequences may be specified by the user for NO, NO ₂ or O ₃ output, with up to 15 steps for each sequence from 0 to 500 ppb for O ₃ and NO ₂ (1000 ppb for NO); stored in internal memory |
| Transportation and Temperature Storage Ranges | –20°C to 60°C |

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Theory of Operation



The Model 714 NO₂/NO/O₃ Calibration Source combines the 2B Technologies Model 306 Ozone Calibration SourceTM and the Model 408 Nitric Oxide Calibration SourceTM to produce calibrated sources of O₃, NO and NO₂. The Model 714 makes use of two low-pressure mercury lamps, one to photolyze oxygen to produce O₃, and a second lamp to photolyze nitrous oxide (N₂O) to produce NO. Nitrogen dioxide is produced in the stoichiometric reaction of O₃ with excess NO to convert NO to NO₂. A schematic of the Model 714 is shown below:

For the production of ozone, a small flow of scrubbed and dried ambient air is directed into the O_3 photolysis chamber (lower part of the diagram). The vacuum UV emission lines of a low-pressure mercury lamp near 185 nm are absorbed by O_2 to produce oxygen atoms, and the oxygen atoms rapidly attach to O_2 via a collisionally stabilized reaction to form O_3 . The well-known mechanism, which also is responsible for the presence of Earth's protective ozone layer, is as follows:

$$O_2 + h\nu \rightarrow 0 + 0 \tag{1}$$

$$2 \left[\mathsf{O} + \mathsf{O}_2 + \mathsf{M} \rightarrow \mathsf{O}_3 + \mathsf{M} \right] \tag{2}$$

Net:
$$3 O_2 + h\nu \rightarrow 2 O_3$$
 (3)

where hv symbolizes a photon of light and M is any molecule (e.g., N₂, O₂, H₂O, Ar) that removes the energy released in reaction (2). The concentration of O₃ produced in a flowing stream of air depends on the intensity of the photolysis lamp, the concentration of oxygen (determined by pressure and temperature), and the residence time in the photolysis cell (determined by volumetric flow rate and cell volume). By holding these parameters constant, it is possible to produce a flow of air containing a constant concentration of O₃, and the concentration of O₃ produced can be varied most conveniently by varying the lamp intensity. An in-line dryer reduces the humidity of the air stream and enables more precise control of ozone produced in the photolysis chamber.

For the production of NO, a mass flow controller (MFC) directs a ~40 cc/min flow of nitrous oxide from an N₂O cartridge or gas cylinder into the N₂O photolysis chamber (upper part of diagram). The vacuum UV emission lines of a low-pressure mercury lamp near 185 nm are absorbed by N₂O to produce electronically excited oxygen atoms, O ($^{1}D_{2}$). A large fraction (~59%) of these highly energetic oxygen atoms react with N₂O to form NO:

$$N_2O + h\nu \rightarrow N_2 + O(^1D_2) \tag{4}$$

$$O(^{1}D_{2}) + N_{2}O \rightarrow 2 NO$$
(5)

Net:
$$2 N_2 O + h\nu \rightarrow N_2 + 2 NO$$
 (6)

where the term hv represents a photon of light. As for ozone, and the concentration of NO produced can be varied most conveniently by varying the lamp intensity. The nitrous oxide can be supplied by a 16-g or 8-g disposable cartridge (e.g., Whip-It or other whipped cream charger) in the Model 714's Portable N₂O Source, thereby eliminating the need for a compressed gas cylinder.

For the production of nitrogen dioxide, a known concentration of O_3 is reacted with a large excess of NO according to the reaction:

$$NO + O_3 \rightarrow NO_2 + O_2 \tag{7}$$

The NO stream combines with the O_3 stream in the Reaction Zone. In order to avoid complications from secondary reactions, nitric oxide is always generated in excess to completely consume all of the O_3 . The concentration of NO₂ produced is equal to the initial ozone concentration. This method of producing known amounts NO₂ equal to the amount of ozone added has long been referred to as the "gas phase titration" (GPT) method. The calibrated NO₂ stream will therefore also contain the excess NO.

By powering the appropriate lamp, one can choose to produce only a calibrated amount of NO, only a calibrated amount of O_3 , or one can power both lamps simultaneously to produce a calibrated amount of NO_2 .

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Features

- Calibrate for NO, NO₂, or O₃ using one instrument
- Low power consumption (~18 watt on average)
- Lightweight (~15.5 lb, 7.0 kg), standard rack-mount size
- Convenient touch screen user interface
- Program up to 99 different sequences, stored in internal memory
- A sequence can have up to 15 steps and can be repeated up to 99 times
- Portable N₂O Source eliminates the need for compressed gas cylinders
- Economical, replacing the use of multiple other commercially available instruments

System Includes

- Model 714 NO₂/NO/O₃ Calibration Source
- Power Adapter (5 amp, 100-240 VAC to 12 VDC) with select power cord
- Patented Portable N₂O Source
- Model 714 Dryer
- External Ozone Scrubber
- Operation Manual on USB Stick
- Calibration Data and NIST-Traceable Calibration Certificate
- Instrument Birth Certificate
- One-Year Warranty



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